Name: \_\_\_\_\_

Honors Chemistry:  $\Box$  yellow  $\ \Box$  blue  $\ \Box$  red

## Intermolecular Forces worksheet

For each pair of compounds:

- Circle the compound of each pair that would have the *lower* melting and boiling point (*weaker* intermolecular forces).
- Give the reason for your choice.
- Describe the types of intermolecular forces that occur in each of the two compounds.
- 1. NaCl HCl
- 2. CaS KCl
- 3.  $H_2O$   $N_2O$
- 4. CO  $CO_2$
- 5. HBr HF
- $6. \quad CH_4 \qquad C_8H_{18}$