

Introduction: Gases

Unit: Gases

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Partial Pressures	141
Molecular Speed & Effusion	148

Standards addressed in this chapter:

Massachusetts Curriculum Frameworks & Science Practices (2016):

HS-PS2-8(MA) Use kinetic molecular theory to compare the strengths of electrostatic forces and the prevalence of interactions that occur between molecules in solids, liquids, and gases. Use the combined gas law to determine changes in pressure, volume, and temperature in gases.

Use this space for summary and/or additional notes: