Introduction: Chemical Reactions

Unit: Chemical Reactions

Details

Big Ideas

Topics covered in this chapter:

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Types of Chemical Reactions	.374
Predicting the Products of Chemical Reactions	.379
Activity (Reactivity) Series	.384
Balancing Chemical Equations	.387
Net Ionic Equations	.395

Standards addressed in this chapter:

Massachusetts Curriculum Frameworks & Science Practices (2016):

- **HS-PS1-2** Use the periodic table model to predict and design simple reactions that result in two main classes of binary compounds, ionic and molecular. Develop an explanation based on given observational data and the electronegativity model about the relative strengths of ionic or covalent bonds.
- **HS-PS1-7** Use mathematical representations and provide experimental evidence to support the claim that atoms, and therefore mass, are conserved during a chemical reaction. Use the mole concept and proportional relationships to evaluate the quantities (masses or moles) of specific reactants needed in order to obtain a specific amount of product.

Use this space for summary and/or additional notes: